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HYDROGEN PEROXIDE

1. PRODUCT AND COMPANY IDENTIFICATION

Product Identifier

Product Name Hydrogen Peroxide 50% Standard

Other means of identification

CAS-No 7722-84-1

Recommended use of the chemical and restrictions on use

Recommended Use: Industrial bleaching, processing, pollution abatement and general oxidation reactions

Restrictions on Use

Manufacturer/Supplier : Novichem Co.

No.30, Magnolia Str, Qaem Maqam Farahani Ave.

Tehran 15886/13941 IRAN

Contact telephone number

(Product information)

: +98-21-88329799 (Hunting)

2. HAZARDS IDENTIFICATION

Classification

OSHA Regulatory Status

This material is considered hazardous by the OSHA Hazard Communication Standard (29 CFR 1910.1200)

Acute toxicity - Oral	Category 4
Acute toxicity - Inhalation (Vapors)	Category 4
Skin corrosion/irritation	Category 1 Sub-category B
Serious eye damage/eye irritation	Category 1
Specific target organ toxicity (single exposure)	Category 3
Oxidizing Liquids	Category 2



GHS Label elements, including precautionary statements

EMERGENCY OVERVIEW

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Danger

Hazard Statements

H314 - Causes severe skin burns and eye damage

H302 - Harmful if swallowed

H332 - Harmful if inhaled

H335 - May cause respiratory irritation

H272 - May intensify fire; oxidizer



Precautionary Statements - Prevention

P271 - Use only outdoors or in a well-ventilated area

P260 - Do not breathe mist, vapours or spray.

P280 - Wear protective gloves/ protective clothing/ eye protection/ face protection

P210 - Keep away from heat/sparks/open flames/hot surfaces. - No smoking

P220 - Keep/Store away from clothing/flammable materials/combustibles

P221 - Take any precaution to avoid mixing with combustibles/flammables

Precautionary Statements - Response

P305 + P351 + P338 - IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing

P310 - Immediately call a POISON CENTER or doctor

P303 + P361 + P353 - IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with water/ shower

P363 - Wash contaminated clothing before reuse

P304 + P340 - IF INHALED: Remove person to fresh air and keep comfortable for breathing

P312 - Call a POISON CENTER or doctor if you feel unwell

P301 + P330 + P331 - IF SWALLOWED: rinse mouth. Do NOT induce vomiting

P310 - Immediately call a POISON CENTER or doctor

P370 + P378 - In case of fire: Use water for extinction

Hazards not otherwise classified (HNOC)

No hazards not otherwise classified were identified.

Other Information

Keep container in a cool place out of direct sunlight. Store only in vented containers. Do not store on wooden pallets. Do not return unused material to its original container. Avoid contamination - Contamination could cause decomposition and generation of oxygen which may result in high pressure and possible container rupture. Empty drums should be triple rinsed with water before discarding.



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3. COMPOSITION/INFORMATION ON INGREDIENTS

Formula H2O2

Chemical name	CAS-No	Weight %
Hydrogen peroxide	7722-84-1	50
Water	7732-18-5	50

Occupational exposure limits, if available, are listed in section 8

4. FIRST AID MEASURES

Eye Contact Rinse immediately with plenty of water, also under the eyelids, for at least 15 minutes.

Remove contact lenses, if present, after the first 5 minutes, then continue rinsing. Seek

immediate medical attention/advice.

Skin ContactTake off contaminated clothing. Rinse skin immediately with plenty of water for 15-20

minutes. Call a poison control center or doctor for further treatment advice.

Inhalation Move to fresh air. If person is not breathing, contact emergency medical services, then give

artificial respiration, preferably mouth-to-mouth if possible. Call a poison control center or

doctor for further treatment advice.

Ingestion Rinse mouth. Do not induce vomiting. If conscious, give 2 glasses of water. Get immediate

medical attention. Never give anything by mouth to an unconscious person.

Most important symptoms and effects, both acute and delayed

Hydrogen Peroxide irritates respiratory system and, if inhaled, may cause inflammation and pulmonary edema. The effects may not be immediate. Overexposure symptoms are coughing, giddiness and sore throat. In case of accidental ingestion, necrosis may result from mucous membrane burns (mouth, esophagus and stomach). Oxygen rapid release may cause stomach swelling and hemorrhaging, which may product major, or even fatal,

injury to organs if a large amount has been ingested.

In case of skin contact, may cause burns, erythema, blisters or even necrosis.

5. FIRE-FIGHTING MEASURES

Suitable Extinguishing Media Water. Do not use any other substance.

Specific Hazards Arising from the

Chemical

In closed unventilated containers, risk of rupture due to the increased pressure from

decomposition.

Flammable properties Contact with combustible material may cause fire

Hazardous Combustion Products On decomposition product releases oxygen which may intensify fire.

Explosion data

Sensitivity to Mechanical Impact Sensitivity to Static Discharge Not sensitive. Not sensitive.

Protective equipment and precautions for firefighters

Use water spray to cool fire exposed surfaces and protect personnel. Move containers from fire area if you can do it without risk. As in any fire, wear self-contained breathing apparatus



6. ACCIDENTAL RELEASE MEASURES

Personal Precautions Avoid contact with skin, eyes and clothing. Wear personal protective equipment. Isolate and

post spill area. Keep people away from and upwind of spill/leak. Eliminate all sources of

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ignition and remove combustible materials.

Other Combustible materials exposed to hydrogen peroxide should be immediately submerged in

or rinsed with large amounts of water to ensure that all hydrogen peroxide is removed. Residual hydrogen peroxide that is allowed to dry (upon evaporation hydrogen peroxide can concentrate) on organic materials such as paper, fabrics, cotton, leather, wood or other

combustibles can cause the material to ignite and result in fire.

Environmental Precautions See Section 12 for additional Ecological Information.

Methods for Containment Dike to collect large liquid spills. Stop leak and contain spill if this can be done safely. Small

spillage: Dilute with large quantities of water.

Methods for cleaning up Flush area with flooding quantities of water. Hydrogen peroxide may be decomposed by

adding sodium metabisulfite or sodium sulfite after diluting to about 5%.

7. HANDLING AND STORAGE

HandlingUse only in well-ventilated areas. Keep/Store away from clothing/ combustible materials.

Wear personal protective equipment. Never return unused hydrogen peroxide to original container. Contamination may cause decomposition and generation of oxygen gas which could result in high pressures and possible container rupture. Empty drums should be triple rinsed with water before discarding. Utensils used for handling hydrogen peroxide should only be made of glass, stainless steel, aluminum or plastic. Pipes and equipment should be passivated before first use. Hydrogen peroxide should be stored only in vented containers

and transferred only in a prescribed manner.

Storage Keep containers in cool areas out of direct sunlight and away from combustibles. Provide

mechanical general and/or local exhaust ventilation to prevent release of vapor or mist into work environment. Containers must be vented. Keep/store only in original container. Store rooms or warehouses should be made of non-combustible materials with impermeable floors. In case of release, spillage should flow to safe area. Containers should be visually inspected on a regular basis to detect any abnormalities (swollen drums, increases in

temperature, etc.).

8. EXPOSURE CONTROLS/PERSONAL PROTECTION

Control parameters

Exposure Guidelines Ingredients with workplace control parameters.

Chemical name	ACGIH TLV	OSHA PEL	NIOSH	Mexico	
Hydrogen peroxide	TWA: 1 ppm	TWA: 1 ppm	IDLH: 75 ppm	Mexico: TWA 1 ppm	
7722-84-1		TWA: 1.4 mg/m ³ TWA: 1 ppm		Mexico: TWA 1.5 mg/m ³	
			TWA: 1.4 mg/m ³	Mexico: STEL 2 ppm	
				Mexico: STEL 3 mg/m ³	
Chemical name	British Columbia	Quebec	Ontario TWAEV	Alberta	
Hydrogen peroxide	TWA: 1 ppm	TWA: 1 ppm	TWA: 1 ppm	TWA: 1 ppm	
7722-84-1		TWA: 1.4 mg/m ³		TWA: 1.4 mg/m ³	



Individual protection measures, such as personal protective equipment

Eye/Face ProtectionUse chemical splash-type monogoggles and a full-face shield made of polycarbonate,

acetate, polycarbonate/acetate, PETG or thermoplastic.

Skin and Body Protection For body protection wear impervious clothing such as an approved splash protective suit

made of SBR rubber, PVC (PVC Outershell w/Polyester Substrate), Gore-Tex (Polyester trilaminate w/Gore-Tex), or a specialized HAZMAT Splash or Protective Suite (Level A, B, or C). For foot protection, wear approved boots made of NBR, PVC, Polyurethane, or neoprene. Overboots made of Latex or PVC, as well as firefighter boots or specialized HAZMAT boots are also permitted. DO NOT wear any form of boot or overboot made of nylon or nylon blends. DO NOT USE cotton, wool or leather as these materials react rapidly with higher concentrations of hydrogen peroxide. Completely submerge hydrogen peroxide contaminated clothing or other materials in water prior to drying. Residual hydrogen peroxide, if allowed to dry on materials such as paper, fabrics, cotton, leather, wood or

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other combustibles, can cause the material to ignite and result in a fire.

Hand Protection For hand protection, wear approved gloves made of nitrile, PVC, or neoprene. DO NOT

use cotton, wool or leather for these materials react RAPIDLY with higher concentrations of hydrogen peroxide. Thoroughly rinse the outside of gloves with water prior to removal.

Inspect regularly for leaks.

Respiratory Protection If concentrations in excess of 10 ppm are expected, use NIOSH/DHHS approved

self-contained breathing apparatus (SCBA) or other approved air-supplied respirator (ASR) equipment (e.g., a full-face airline respirator (ALR)). DO NOT use any form of air-purifying respirator (APR) or filtering facepiece (dust mask), especially those containing oxidizable

sorbants such as activated carbon.

Hygiene measures Avoid breathing vapors, mist or gas. Clean water should be available for washing in case of

eye or skin contamination. .

General information Protective engineering solutions should be implemented and in use before personal

protective equipment is considered.

9. PHYSICAL AND CHEMICAL PROPERTIES

Information on basic physical and chemical properties

Appearance Clear, colorless liquid

Physical State
Color
Colorless
Odor
Odor threshold
pH

Melting point/freezing point

Liquid
Colorless
odorless
Not applicable
<= 3.0
-52 °C

Boiling Point/Range 114 °C
Flash point Not flammable

Evaporation Rate > 1 (n-butyl acetate=1)
Vapor pressure 18 mm Hg @ 30 °C No
Vapor density information available 1.2

Density @ 20 °C Specific gravity 1.2

Water solubility completely soluble
Solubility in other solvents
Partition coefficient coefficient completely soluble
No information available log Kow = -1.5 @ 20 °C

Autoignition temperature Not combustible



100 °C (adiabatic) **Decomposition temperature** 1.17 cP @ 20 °C Viscosity, kinematic Viscosity, dynamic No information available **Explosive properties** No information available Strong oxidizer

Oxidizing properties 34

Molecular weight

Bulk density Not applicable

10. STABILITY AND REACTIVITY

Reactivity Reactive and oxidizing agent.

Chemical Stability Stable under normal conditions. Decomposes on heating. Stable under recommended

storage conditions.

Possibility of Hazardous Reactions Contact with organic substances may cause fire or explosion. Contact with metals, metallic

ions, alkalis, reducing agents and organic matter (such as alcohols or terpenes) may

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produce self-accelerated thermal decomposition.

Hazardous polymerization Hazardous polymerization does not occur.

Conditions to avoid Excessive heat; Contamination; Exposure to UV-rays; pH variations.

Combustible materials. Copper alloys, galvanized iron. Strong reducing agents. Heavy Incompatible materials

metals. Iron. Copper alloys. Contact with metals, metallic ions, alkalis, reducing agents and organic matter (such as alcohols or terpenes) may produce self-accelerated thermal

decomposition.

Hazardous Decomposition Products Oxygen which supports combustion. Liable to produce overpressure in container

11. TOXICOLOGICAL INFORMATION

Serious eye damage/eye irritation

Skin corrosion/irritation

Corrosive. Risk of serious damage to eyes. Corrosive to skin. Causes severe burns.

Sensitization Did not cause sensitization on laboratory animals.

Information on toxicological effects

Symptoms Vapors, mists, or aerosols of hydrogen peroxide can cause upper airway irritation,

inflammation of the nose, hoarseness, shortness of breath, and a sensation of burning or tightness in the chest. Prolonged exposure to concentrated vapor or to dilute solutions can cause irritation and temporary bleaching of skin and hair. Exposure to vapor, mist, or

aerosol can cause stinging pain and tearing of eyes.

Delayed and immediate effects as well as chronic effects from short and long-term exposure

Carcinogenicity This product contains hydrogen peroxide. The International Agency for Research on

Cancer (IARC) has concluded that there is inadequate evidence for carcinogenicity of hydrogen peroxide in humans, but limited evidence in experimental animals (Group 3 - not

classifiable as to its carcinogenicity to humans). The American Conference of



Mutagenicity This product is not recognized as mutagenic by Research Agencies

In vivo tests did not show mutagenic effects

Reproductive toxicity This product is not recognized as reprotox by Research Agencies. No toxicity to

reproduction in animal studies.

STOT - single exposure May cause respiratory irritation.

STOT - repeated exposure Not classified.

Target organ effects Eyes, Respiratory System, Skin.

Aspiration hazard No information available.

12. ECOLOGICAL INFORMATION

Ecotoxicity

Ecotoxicity effects Hydrogen peroxide is naturally produced by sunlight (between 0.1 and 4 ppb in air and

0.001 to 0.1 mg/L in water). Not expected to have significant environmental effects.

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Persistence and degradability Hydrogen peroxide in the aquatic environment is subject to various reduction or oxidation

processes and decomposes into water and oxygen. Hydrogen peroxide half-life in freshwater ranged from 8 hours to 20 days, in air from 10 - 20 hours, and in soils from minutes to hours depending upon microbiological activity and metal contamination.

Bioaccumulation Material may have some potential to bioaccumulate but will likely degrade in most

environments before accumulation can occur.

Mobility Will likely be mobile in the environment due to its water solubility but will likely degrade over

time.

Other Adverse Effects Decomposes into oxygen and water. No adverse effects.

13. DISPOSAL CONSIDERATIONS

Waste disposal methods Dispose of in accordance with local regulations. Can be disposed as waste water, when in

compliance with local regulations.

US EPA Waste Number D001; D002.

Contaminated Packaging Dispose of in accordance with local regulations.



14. TRANSPORT INFORMATION

DOT

UN/ID no UN 2014

Proper Shipping Name HYDROGEN PEROXIDE, AQUEOUS SOLUTION

Hazard class 5.
Subsidiary class 8
Packing Group II

TDG

UN/ID no UN 2014

Proper Shipping Name HYDROGEN PEROXIDE, AQUEOUS SOLUTION

Hazard class 5.1 Subsidiary class 8 Packing Group II

ICAO/IATA Hydrogen peroxide (>40%) is forbidden on Passenger and Cargo Aircraft. Air regulation

permit shipment of Hydrogen Peroxide (<=40%) in non-vented containers for Air Cargo Only aircraft, as well as for Passenger and Cargo aircraft. HOWEVER, all Hydrogen Peroxide containers are vented and therefore, air shipments of H2O2 are not permitted. IATA air regulations state that venting of packages containing oxidizing substances is not

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permitted for air transport.

IMDG/IMO

UN/ID no UN 2014

Proper Shipping Name HYDROGEN PEROXIDE, AQUEOUS SOLUTION

Hazard class 5.1 Subsidiary Hazard Class 8 Packing Group II

OTHER INFORMATION Protect from physical damage. Keep drums in upright position. Drums should not be

stacked in transit. Do not store drums on wooden pallets.

15. REGULATORY INFORMATION

U.S. Federal Regulations

SARA 313

Section 313 of Title III of the Superfund Amendments and Reauthorization Act of 1986 (SARA). This product does not contain any chemicals which are subject to the reporting requirements of the Act and Title 40 of the Code of Federal Regulations, Part 372

SARA 311/312 Hazard Categories

This product has the following hazards that are reportable under The Emergency Planning and Community Right-to-Know rule (EPCRA Tier II)

Clean Water Act

This product does not contain any substances regulated as pollutants pursuant to the Clean Water Act (40 CFR 122.21 and 40 CFR 122.42)



CANADA

Environmental Emergencies

Component	Canada - Environmental	Canada - Environmental	Canada - Environmental	Canada - Environmental	
	Emergencies - Part 1	Emergencies - Part 1	Emergencies - Part 2	Emergencies - Part 2	
	Substances -	Substances -	Substances -	Substances -	
	Substances Likely to	Substances Likely to	Substances Hazardous	Substances Hazardous	
	Explode - Minimum	Explode - Minimum	When Inhaled - Minimum	When Inhaled - Minimum	
	Threshold Quantities	Mixture Concentrations	Threshold Quantities	Mixture Concentrations	
Hydrogen peroxide	3.40 tonnes Minimum	52			
7722-84-1 (50)	quantity				

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Canadian National Pollutant Release Inventory

This product contains no substances reportable under Canada's National Pollutant Release Inventory regulations.

International Inventories

Component	TSCA (United States)	DSL (Canada)	EINECS/EL INCS (Europe)	ENCS (Japan)	China (IECSC)	KECL (Korea)	PICCS (Philippines)	AICS (Australia)	NZIoC (New Zealand)
Hydrogen peroxide 7722-84-1 (50)	Χ	Х	X	X	Х	X	X	Х	Х

16. OTHER INFORMATION

Information contained herein is provided in good faith but makes no representation as to its comprehensiveness or accuracy. This document is intended only as a guide to the appropriate precautionary handling of the material by a properly trained person using this product. Individuals receiving the information must exercise their independent judgment in determining its appropriateness for a particular purpose.